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work = output work $d_i \times F_i = d_o \times F_o$ (1 m)(50 N) = $d_o \times (1000$
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Stephen Murray Thermodynamics Answers

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suggested for APB) Thermodynamic Processes Vocabulary (M) - basic definitions. Thermodynamic Processes (MC) - learn the differences in detail. * Heat, Temperature, Internal Energy and Thermodynamic Process (MC) Qc or Qh? (MC) - Will help you work the cyclic process questions.

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Murraydone by the gas. $\Delta U = Q - W = (0.0) - W = -$

W . The decrease of the internal energy of the gas results in

a temperature decrease of the gas. and a final temperature

less than 30 C. If the gas were air : $T_2 = (T_1) (P_2/P_1)^{\gamma}$.

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Heat And Thermo 1 Answer Question 7 An ideal gas is taken

through a cyclic thermodynamics process through four

steps. The amount of heat involved in the steps are $Q_1 =$

5960 J , $Q_2 = -5600 \text{ J}$, $Q_3 = -3000 \text{ J}$, $Q_4 = -3600 \text{ J}$ respectively.

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gas is taken through a cyclic thermodynamics process
through four steps. The amount of heat involved in the
steps are $Q_1 = 5960 \text{ J}$, $Q_2 = -5600 \text{ J}$, $Q_3 = -3000 \text{ J}$, $Q_4 =$
 -3600 J respectively.

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