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The elements
carbon, hydrogen,
nitrogen, oxygen,
sulfur, and
phosphorus are the
key building blocks
of the chemicals
found in living
things. They form
the carbohydrates,
nucleic acids,
proteins, and lipids
(all of which will be

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life (defined later in this chapter) that are the fundamental molecular components of all organisms. In this chapter, we will discuss these important building blocks and learn how the unique properties of the atoms of different elements affect

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~~Chemistry of Life ...~~
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crossword puzzle
answer key.
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The chemistry of life begins with the basic principles of bond formation and bond breaking, and the nature of the different compounds formed. Life revolves around the balancing act between the energy released as bonds are broken

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Life and the energy
Key
taken in as bonds
Chapter 24
are formed. Life on
earth depends on
the nature of the
carbon atom and
the nature of
water.

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For example, in Key
Biology Chapter 2
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the Chemistry of
Life, students learn
about carbon,
nitrogen, oxygen,
hydrogen, and
iodine. The
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teaches students
the correct and
incorrect ways of
using the scales to
solve problems and

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However, as stated before, the online students will not be lectured about biology.

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SAMPLE ANSWER:
Carbon is the
primary element

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things. SAMPLE
ANSWER:

Organisms use
carbon compounds
to form four types
of molecules:

lipids,
carbohydrates,
nucleic acids, and
proteins. SAMPLE
ANSWER: A lot of
what that happens
in an organism is

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life based on chemical
reactions. SAMPLE
ANSWER: Enzymes
are proteins that
speed up

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Answers.

Worksheet

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Worksheet Answers

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- Your job will include things like quality control dosimetry, therapy planning and other tasks that are essential, but your occupation environment might be different, based on the place you work. Your capacity

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to earn is based on
the amount of work
you would like to
put in.

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The four main
types of carbon-
based molecules in
organisms are
carbohydrates,
lipids, nucleic

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acids, and.
starches. fatty
acids.

monosaccharides.
proteins. Q. Both
animal fats and
plant oils are made
up of glycerol and.
phospholipids. fatty
acids.

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of Life | Other Quiz
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(pages 35-39) This
section identifies
the three particles
that make up
atoms. It also
explains how
atoms of the same
element can have

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a different number
of neutrons and
describes the two
main types of
chemical bonds.
Atoms (page 35) 1.

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Question: The
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EXERCISE 1:

TESTING FOR
BIOMOLECULES

Data Sheet Table

1. Testing For
Proteins Results

Sample Initial Color
Hypothesis (Protein

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+/-) Final Answer Key

Protein Present?

Albumin (1) Gelatin

(2) Glucose (3)

Water (4) E Table

2.

~~Solved: The~~

~~Chemistry Of Life~~

~~EXERCISE 1:~~

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