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only change the
value of
coefficient in
front of the
element or
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not the
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containing
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balanced or not,
and how that
relates to the
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mass. Each
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equation is

mostly a matter

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equation
described in
section 4.1 is
balanced,
meaning that
equal numbers of

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atoms for each element involved in the reaction are represented on the reactant and product sides. This is a requirement the equation must satisfy to be consistent with the law of conservation of matter.

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questions are
relatively easy

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