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Total work =100.9 kJ. answer: P=0.3 atm. 2.6 Ten moles of ideal gas, in the initial state $P_1 = 10$ atm, $T_1 = 300$ K, are taken round the following cycle: a. A reversible change of state along a straight line path on the P-V diagram to the state $P = 1$ atm, $T = 300$ K, b. A reversible isobaric compression to $V = 24.6$ liters, and c.

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