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2 Measurements and Calculations

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2 does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl₂ is an ionic solid. In the crystal form, its ions are locked in position. Dissolving the crystal in water releases the ions to move freely, allowing them to conduct electricity. 2. Explain the following statements at the molecular level: a.

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4798 CHAP 9 REVIEW - srvhs.org. CHAPTER 9 REVIEW Stoichiometry SECTION 9-3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

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