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Chapter 2 Beginning

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A covalent bond in  
which two atoms share  
three pairs of electrons is



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called a(n) bond. 6. A bond in which two atoms share one pair of electrons is a bond. 7. A nonbonding pair of electrons in the valence shell of an atom is called a(n) pair. Answer the following items in the space provided. 8.

Complete each statement below by choosing a term from

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structures 2. b A

covalent bond consists  
of (a) a shared electron.

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To Chemical Bonding  
Answer Key

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SHORT ANSWER

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Answer the following questions in the space provided. 1.

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valence electrons are attracted by the positively charged nuclei of two atoms and thus are shared between both atoms. 2. The  $H_2$  molecule is stable because each hydrogen atom now has a shared pair of electrons and has achieved a stable noble gas configuration. 3.

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Answer Key Chapter 12

Summary.  $2 \times$  atomic  
mass of carbon =

$2 \text{ atoms}(12.011 \text{ amu}$

$\text{atoms}) = 24.022 \text{ amu. } 6$

$\times$  atomic mass of

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hydrogen =  
2atoms(1.0079amu  
atoms) = 6.0474amu. 1  
× atomic mass of  
oxygen =  
1atoms(15.9994amu  
atoms) = 15.994amu.

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...  
each statement or best  
answers each question.

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1. Chemical bonding in metals is a. the same as ionic bonding. b. the same as covalent bonding. c. a combination of ionic and covalent bonding. d. different from ionic or covalent bonding. 2. The valence electrons in a metallic bond move freely throughout the network of metal atoms.

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A covalent bond forms when two or more valence electrons are attracted Holt Chemistry Concept Review Holt Chemistry ... Holt Chemistry Concept Review - trumpetmaster.com a concept of chemical bonding theory that is based on the assumption

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that atoms tend to have either empty valence shells or full valence shells of eight electrons.

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Review

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