

## Chemistry Solutions

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Solution, in chemistry, a homogenous mixture of two or more substances in relative amounts that can be varied continuously up to what is called the limit of solubility. The term solution is commonly applied to the liquid state of matter, but solutions of gases and solids are possible.

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A solution is defined as a homogenous mixture which mainly comprises of two components namely solute and solvent. For example, salt and sugar is a good illustration of a solution. A solution can be categorized into several components.

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A solution is a homogeneous mixture of two or more substances. A solution may exist in any phase . A solution consists of a solute and a solvent. The solute is the substance that is dissolved in the solvent. The amount of solute that can be dissolved in solvent is called its solubility.

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In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent.The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in interactions that are specific to solvation.

[Solution - Wikipedia](#)

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The substance that does the dissolving in a solution. The substance that is being dissolved in a solution. The mixing of different substances. The process in which neutral molecules lose or gain electrons

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