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Solutions CHAPTER

12 REVIEW Solutions

SECTION 2 SHORT

ANSWER Answer the

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following questions in the space provided. 1. The following are statements about the dissolving process.

Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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~~CHAPTER 12~~

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A homogeneous mixture that contains particles in a dispersed phase that do not settle out is a solution. The solution process with water as the solvent. hydration. Liquid solutes and solvents that are not soluble in each other

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are immiscible. A solute molecule that is surrounded by solvent molecules is solvated.

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review. STUDY.
PLAY. solution.
homogeneous mixture
of two or more

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substances,
consisting of ions or
molecules. solute. gas
or solid dissolved in a
liquid or solution
component in small
amount. solvent.
liquid that dissolves a
gas or solid or the
solution component in
greater amount.

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CHAPTER 12

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SECTION 1 SHORT

ANSWER Answer the

following questions in
the space provided. 1.

Match the type of
mixture on the left to
its representative
particle diameter on
the right. c solutions
(a) larger than 1000
nm a suspensions (b)

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1 nm to 1000 nm b
colloids (c) smaller
than 1 nm 2. Identify
the solvent in each of
the following
examples:

~~mc06se cFMSr i vi~~

Chapter 12 Review

Solutions 1) 56 2) = 4,

= 10, = 16, = 22, = 28

3) = 4, = 8, = , = 64, =

4) $an = 2n - 3$ 5) = ?

$4n$ 6) = 5, = 3, = 1, =

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1) 3 + 4 + 5 + ... +
(n + 2) 8) 9) 15 10) d
= -5; = 4, = -1, = -6, =
-11 11) = 94 - 10n; =
14 12) 220 13) = 73, d
= -5, = - 5 14) 34980
15) 25

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~~Solutions , = 10, = 16,
= 22, = 28 + 1) = 2.~~

CHAPTER . 12 .

REVIEW "-- Solutions

. SHORT ANSWER

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Answer the following questions in the space provided. 1.

Describe the errors made by the following students in making molar solutions. 3.

James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

James made several errors.

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Solutions

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CHAPTER 12

REVIEW Solutions

SECTION 3 SHORT

ANSWER Answer the

following questions in
the space provided. 1.

Describe the errors
made by the following
students in making
molar solutions. a.

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James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

b. Mary needs a 0.02 M solution of NaNO_3 . She calculates that she needs 2.00

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is suitable for 9th -

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colloids.

Solutions

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Problems

INSTRUCTIONS:

Write your answers
on lined or blank
sheets of paper. You
do not need to write
the question, ONLY
WRITE THE

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PROBLEM NUMBER
and ANSWERS/SOL
UTIONS. • For
problems that involve
calculations, you must
show your work to get
full credit.

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Section 1- Types of
Mixtures Solutions are
homogeneous
mixtures of two or
more substances in a
single phase. Soluble
describes a
substance as capable
of being dissolved.
Solvent is the
dissolving medium in
a solution.

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of being dissolved.

Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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